

**PCT**WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau

## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification <sup>6</sup> : <b>C01B 31/02, H01M 4/04</b>	<b>A1</b>	(11) International Publication Number: <b>WO 95/06002</b> (43) International Publication Date: <b>2 March 1995 (02.03.95)</b>
<p>(21) International Application Number: <b>PCT/US94/09547</b></p> <p>(22) International Filing Date: <b>23 August 1994 (23.08.94)</b></p> <p>(30) Priority Data: <b>08/110,003</b> <b>23 August 1993 (23.08.93)</b> <b>US</b></p> <p>(71) Applicant: <b>THE REGENTS OF THE UNIVERSITY OF CALIFORNIA [US/US]; 22nd floor, 300 Lakeside Drive, Oakland, CA 94612 (US).</b></p> <p>(72) Inventors: <b>PEKALA, Richard, W.; 802 Cliffside Drive, Pleasant Hill, CA 94523 (US). MAYER, Steven, T.; 16026 Selborne Drive, San Leandro, CA 94578 (US). KASCHMITTER, James, L.; 6291 Alisal Street, Pleasanton, CA 94566 (US). MORRISON, Robert, L.; 1117 Springcreek Drive, Modesto, CA 95355 (US).</b></p> <p>(74) Agent: <b>SARTORIO, Henry, P.; P.O. Box 808, Livermore, CA 94550 (US).</b></p>		<p>(81) Designated States: <b>CA, JP, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE).</b></p> <p><b>Published</b> <i>With international search report.</i></p>
<p>(54) Title: <b>METHOD FOR MAKING THIN CARBON FOAM ELECTRODES</b></p> <p>(57) Abstract</p> <p>A method for fabricating thin, flat carbon electrodes by infiltrating highly porous carbon papers, membranes, felts, metal fibers/powders, or fabrics with an appropriate carbon foam precursor material. The infiltrated carbon paper, for example, is then cured to form a gel-saturated carbon paper, which is subsequently dried and pyrolyzed to form a thin sheet of porous carbon. The material readily stays flat and flexible during curing and pyrolyzing to form thin sheets. Precursor materials include polyacrylonitrile (PAN), polymethylacrylonitrile (PMAN), resorcinol/formaldehyde, catechol/formaldehyde, phenol/formaldehyde, etc., or mixtures thereof. These thin films are ideal for use as high power and energy electrodes in batteries, capacitors, and fuel cells, and are potentially useful for capacitive deionization, filtration and catalysis.</p>		

BEST AVAILABLE COPY

**FOR THE PURPOSES OF INFORMATION ONLY**

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GB	United Kingdom	MR	Mauritania
AU	Australia	GE	Georgia	MW	Malawi
BB	Barbados	GN	Guinea	NE	Niger
BE	Belgium	GR	Greece	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IE	Ireland	NZ	New Zealand
BJ	Benin	IT	Italy	PL	Poland
BR	Brazil	JP	Japan	PT	Portugal
BY	Belarus	KE	Kenya	RO	Romania
CA	Canada	KG	Kyrgyzstan	RU	Russian Federation
CF	Central African Republic	KP	Democratic People's Republic of Korea	SD	Sudan
CG	Congo	KR	Republic of Korea	SE	Sweden
CH	Switzerland	KZ	Kazakhstan	SI	Slovenia
CI	Côte d'Ivoire	LI	Liechtenstein	SK	Slovakia
CM	Cameroon	LK	Sri Lanka	SN	Senegal
CN	China	LU	Luxembourg	TD	Chad
CS	Czechoslovakia	LV	Latvia	TG	Togo
CZ	Czech Republic	MC	Monaco	TJ	Tajikistan
DE	Germany	MD	Republic of Moldova	TT	Trinidad and Tobago
DK	Denmark	MG	Madagascar	UA	Ukraine
ES	Spain	ML	Mali	US	United States of America
FI	Finland	MN	Mongolia	UZ	Uzbekistan
FR	France			VN	Viet Nam
GA	Gabon				

### METHOD FOR MAKING THIN CARBON FOAM ELECTRODES

The United States Government has rights in this invention pursuant to Contract No. W-7405-ENG-48 between the United States Department of Energy and the University of California for the operation of Lawrence Livermore National Laboratory.

### BACKGROUND OF THE INVENTION

The present invention relates to carbon foam hybrid/composite electrodes, particularly to thin film carbon hybrid/composite electrodes, and more particularly to a method for forming thin, flat porous carbon sheets by infiltrating a highly porous fiber matrix with a carbon foam precursor material, curing to form a polymer-saturated material, and pyrolyzing to form a thin sheet of carbon, for use as high density and/or high energy density power electrodes.

Carbon aerogels or foams produced by pyrolyzing organic precursors are known in the art, as exemplified by U.S. Patents No. 4,873,218 issued October 10, 1989; No. 4,997,804 issued March 5, 1991; and No. 5,086,085 issued February 4, 1992, each to R.W. Pekala. Also, recent efforts have been directed to the use of monolithic carbon aerogel electrodes for use in capacitors, etc., as exemplified by U.S. Application Serial No. 07/822,438 filed January 17, 1992, entitled "Supercapacitors Based On Carbon Foams", in the name of J.L. Kaschmitter et al.; and U.S. Application Serial No. 08/(IL-9154), filed \_\_\_\_\_, 1993 entitled "Cell Separator For Use In Bipolar-Stack Energy Storage Devices", in the name of S.T. Mayer et al.

Thin films of porous carbon are desired for use as electrodes in batteries and capacitors because they exhibit lower resistance and can provide high efficiency and high power capability. Thinner films also provide short electrolyte diffusion distances and better access to the carbon. Thin layers or films (<20 mils) of the prior monolithic carbon foams can be difficult to form and handle because they often are brittle.

- 2 -

Processing of monolithic thin films is difficult due to the large amount of shrinkage that accompanies the curing and pyrolyzing, sometimes resulting in cracking of the material as it shrinks. Thus, there is a need for a method to form thin carbon foam hybrid/composite electrodes which overcome the disadvantages of the prior known monolithic carbon foams for use as electrodes.

The present invention satisfies this need by providing a method for producing thin, flat porous carbon electrodes with an electrically continuous, low resistance structure. The method involves the infiltration of a highly porous, low density ( $<0.2\text{g/cc}$ ) material, such as carbon paper or felt, with the appropriate carbon foam precursor material prior to pyrolysis. The infiltrated highly porous material (referred to hereinafter as the "substrate") allows for the carbon foam to shrink around fibers of the material, rather than around itself in the formation of a monolithic structure, thereby eliminating the above-mentioned shrinkage problem of the prior known monolithic carbon foams. The use of the porous substrate material can also allow for roll-to-roll processing of thin sheets of the carbon foam for improved manufacturability.

#### SUMMARY OF THE INVENTION

It is an object of the present invention to provide a method for producing thin carbon foam.

A further object of the invention is to provide thin carbon foam for use as electrodes in energy storage devices.

A further object of the invention is to provide a method for fabricating thin, flat carbon electrodes by infiltrating a highly porous substrate material with an appropriate carbon foam precursor material prior to pyrolysis thereof.

Another object of the invention is to provide a thin film of carbon for use as high power electrodes in capacitors, batteries and fuel cells, for example, with improved flexibility when compared to the more brittle prior monolithic structures.

Another object of the invention is to provide a method which infiltrates highly porous carbon papers with an appropriate carbon foam precursor material, curing the infiltrated material to form a gel-saturated carbon paper, and subsequently pyrolyzing the material to form a thin sheet of carbon foam.

- 3 -

Other objects and advantages of the invention will become apparent from the following description. Basically, the invention involves a method for fabricating thin, flat carbon foam which can be used as electrodes in energy storage devices, such as in a rechargeable lithium battery, as well as for capacitive deionization, filtration and catalysis. The method is basically carried out by infiltrating highly porous material, such as carbon papers, with an appropriate carbon foam precursor material which is subsequently cured and fired at high temperatures (500-3000°C) to form a thin sheet of carbon foam. The material stays flat during curing and firing to form the thin sheets. The thin carbon foam can also, for example, be used in a spirally-wound or "jelly-roll" battery, for example, since the infiltrated carbon foam can be formed to shape more easily than monolithic carbon foam. The density of the thin carbon foam is controlled by the fabrication process and the choice of substrate. Also, sheets can be used in flat, prismatic cells and battery/capacitor designs and are potentially useful for capacitive deionization, filtration and catalysis.

#### DETAILED DESCRIPTION OF THE INVENTION

The invention involves the fabrication of flat, thin (thickness of 1 to 40 mil) carbon foam hybrid/composite electrodes which overcomes the disadvantages of the prior known carbon foam electrodes, such as described and claimed in the two above-referenced Applications S.N. 07/822,438 and S.N. 08/(IL-9154). The fabrication method involves infiltrating highly porous materials, such as carbon or metal fiber paper, membranes, felts, fabrics or metal powders and fibers with an appropriate carbon foam precursor material, the infiltrated material is then cured to form a gel-saturated material, which is subsequently fired in an inert atmosphere at high temperature (1050°C, for example) to form a flat, thin sheet of carbon foam. The infiltrated highly porous material allows the carbon foam to stick around the substrate fibers and expand/contract with the substrate during processing (through formation of microcracks), rather than as a monolithic structure, thereby eliminating the brittle structure of the prior carbon foams. The use of the infiltrated highly porous material can also allow for roll-to-roll processing of the material for improved manufacturability.

- 4 -

Carbon foam precursor materials are well known, as exemplified in the above-referenced U.S. Patents and copending applications. Potential precursor materials for this invention include but are not limited to polyacrylonitrile (PAN), polymethylacrylonitrile (PMAN), resorcinol/formaldehyde, catechol/formaldehyde, phenol/formaldehyde, etc., or mixtures thereof.

The present invention provides a method by which thin film fabrication of carbon foam hybrids/composites can be accomplished by infiltration of highly porous substrate material with a precursor sol or polymer solution. Experimental verification of the invention has been carried out using carbon papers as the porous material and PAN or resorcinol-formaldehyde as the precursor material, as is described as follows:

Example 1:

1. Dissolved approximately 12% w/v polyacrylonitrile (PAN) at an elevated temperature (~150°C) in a 97:3 dimethyl sulfoxide (DMSO): water mixture. Other solvents are described in the above-referenced patents and copending applications.
2. The hot polymer solution was then uniformly infiltrated into carbon paper (125 micron thick, Textron C.05 made by Textron Specialty Chemicals, Lowell, MA; density of ~0.15g/cc), until the paper was saturated. A 3 inch by 3 inch sheet of carbon paper was used, and the mixture was applied on the carbon paper by wicking the polymer solution into the paper.
3. The PAN-saturated carbon paper was then frozen at between -5°C and -150°C, depending on the desired material characteristics.
4. The frozen, PAN-saturated carbon paper, was then freeze-dried at about -6°C with pressure ramping from about 300 milli Torr to 0 milli Torr, for a time period of 8 to 48 hours, which depends on the thickness.
5. The freeze dried carbon paper is then fired in an inert atmosphere furnace to about 1050°C for a time period of 4 hours, resulting in a flat, thin carbon foam (with a thickness of ~125µm). Different firing (pyrolyzing) temperatures (500-3000°C), and different time periods (1 to 8 hours) can be selected depending on the desired

- 5 -

material characteristics. The inert atmosphere may be nitrogen, helium, argon, neon, or other inert gas.

If it is desirable to "activate" the carbon electrode,  $\text{CO}_2$ , steam, or air can be used as the purge gas during pyrolyzing. The freeze-dried carbon paper may be pyrolyzed at temperatures of 800-1200°C in an oxidizing atmosphere (e.g. steam, air,  $\text{CO}_2$ ), to produce "activated" carbon foam, which has a modified pore size distribution and higher surface area than when pyrolyzed in an inert atmosphere.

The density of the above fabricated thin carbon foam was ~0.27g/cc. Higher densities can be produced by either re-saturating the PAN-saturated carbon paper after curing (freezing and freeze-drying), or after pyrolysis and then pyrolyzing again.

Example 2: (Same as 1 but doped with lithium, phosphorous, aluminum, boron or arsenic compounds for improved electrochemical performance). For example, the freeze-dried PAN-carbon paper is exposed to a dilute solution of  $\text{H}_3\text{PO}_4$  in methanol (9.64g per 300cc). The PAN-carbon paper is left in the  $\text{H}_3\text{PO}_4$ /methanol solution for a time period ranging from 8-48 hours. Upon removal from the solution, the methanol is allowed to evaporate leaving a coating of  $\text{H}_3\text{PO}_4$  throughout the PAN/carbon paper. This material is then pyrolyzed as described above. The resultant thin carbon foam shows enhanced intercalation in a lithium-ion battery.

Example 3: Textron C.05 carbon paper, density of ~0.15g/cc, is infiltrated with a resorcinol-formaldehyde (RF) sol formula by reacting 12.35g resorcinol powder, 17.91g 37% formaldehyde, 0.0g deionized  $\text{H}_2\text{O}$ , and 1.12g 0.5M  $\text{Na}_2\text{CO}_3$ , for 10 minutes at ~50°C. The RF saturated carbon paper is placed between 2 glass plates that are then covered and sealed with a polymer film (e.g. Saran Wrap, made by The Dow Chemical Company). The RF/carbon paper is then cured for 1-4 hours at 50°C followed by 1-3 days at 85-95°C. Upon removal from the glass plates, the  $\text{H}_2\text{O}$  in the pores of the RF/carbon paper is exchanged with acetone. The acetone is then evaporated to form an air-dried RF/carbon paper. This material is pyrolyzed at 1050°C for 4 hours in a nitrogen atmosphere. The resultant thin film carbon electrode has a density of ~0.6g/cc. Density of the final material can be controlled by the RF sol concentration or multiple infusions. The air drying technique is described and claimed in copending U.S. Application Serial No.

- 6 -

08/041,503 filed April 1, 1993, entitled "Method Of Low Pressure And/Or Evaporative Drying of Aerogel".

Example 4: Same as described above in Example 3, but using subcritical or supercritical drying using carbon dioxide, as taught in the above-referenced U.S. Patents to Pekala.

Example 5: Chemical activation of the carbon electrodes by exposure to an 8M HNO<sub>3</sub> solution for 24-28 hours at 50°C. Provides improved double/layer capacitance results.

Example 6: (Activated Example 4, CO<sub>2</sub> as the purge gas for 3 hours at 800-1200°C.

Example 7: Same as Example 1, but using 3 mil AFS (Advanced Fiber Substrate) Nickel Metal Fabric, made by National Standard Company, Corbin, Tenn. as the substrate to be impregnated. Other metal fibers, powders, or fabrics containing nickel, titanium, aluminum, or stainless steel, and mixtures thereof may be used.

The freeze-dried carbon paper may be pyrolyzed at temperatures of 600-1200°C in an oxidizing atmosphere (e.g. steam, air, CO<sub>2</sub>), to produce "activated" carbon foam, which has a different pore size distribution and higher surface area than when pyrolyzed in an inert atmosphere.

The thus produced flat, thin films of carbon foam are ideal for high power electrodes in energy storage devices, such as a lithium intercalation anode in a rechargeable lithium ion battery, or as electrodes in capacitors, fuel cells, etc. These flat, thin carbon foams also have potential use in capacitive deionization, filtration and catalysis. In addition, the thin, flat carbon foams can be used in a spirally-wound or "jelly-roll" battery, since the thin carbon foam can be easily shaped.

It has thus been shown that the present invention provides a fabrication process for producing flat, thin carbon foam hybrid/composite electrodes. This is accomplished by the infiltration of a highly porous material, such as carbon paper, into the carbon foam during fabrication of the carbon foam.

While an example of a specific sequence of operational steps, materials, parameters, temperatures, uses, etc., have been set forth to describe the principles of the fabrication process of this invention, such is not intended to limit the invention to the specifics described. Modifications and changes will become apparent to those skilled in the



- 7 -

art, and it is intended that the invention be limited only by the scope of the appended claims.

CLAIMS

1. A method for making thin carbon foam, comprising the steps of:  
infiltrating a highly porous material substrate with a carbon foam precursor material;  
curing the thus infiltrated and precursor materials; and  
pyrolyzing the thus cured materials.
2. The method of Claim 1, additionally including the step of forming the carbon foam precursor material.
3. The method of Claim 2, wherein the precursor material is a precursor solution formed from a precursor, a solvent and water.
4. The method of Claim 3, wherein the precursor solution is formed by dissolving about 12% PAN in a 97:3 DMSO:H<sub>2</sub>O mixture.
5. The method of Claim 4, additionally including the step of applying the mixture to the highly porous material until the material is saturated.
6. The method of Claim 3, wherein the precursor solution is formed by reacting 12.35g resorcinol powder, 17.91g 37% formaldehyde, 0.0g deionized H<sub>2</sub>O, and 1.12g 0.5 Na<sub>2</sub>CO<sub>3</sub>.
7. The method of Claim 6, additionally including the step of applying the solution to the highly porous material until the material is saturated.
8. The method of Claim 1, wherein the highly porous material is selected from the group consisting of carbon paper, felt and cloth.

9. The method of Claim 1, wherein the highly porous material is selected from the group consisting of a matrix of small diameter fibers.

10. The method of Claim 1, wherein the step of curing is carried out by freezing the saturated highly porous material, and then freeze-drying the frozen material.

11. The method of Claim 1, wherein the step of pyrolyzing the thus cured material is carried out by firing in an inert atmosphere furnace at a temperature of 500-3000°C.

12. The method of Claim 11, wherein the step of pyrolyzing is carried out at a temperature of about 1050°C, and wherein the inert atmosphere is nitrogen.

13. The method of Claim 11, wherein the step of pyrolyzing is carried out in an oxidizing atmosphere and a temperature of about 600-1200°C.

14. The method of Claim 1, wherein the highly porous material is selected from the group of carbon papers, membranes, felts, metal fibers and fabrics.

15. The method of Claim 1, wherein the carbon foam precursor material includes material selected from the group consisting of polyacrylonitrile (PAN), polymethylacrylonitrile (PMAN), phloroglucinal/formaldehyde, resorcinol/formaldehyde, catechol/formaldehyde, phenol/formaldehyde, and mixtures thereof.

16. The method of Claim 1, wherein the high porous material is carbon paper, and wherein the carbon foam precursor material is composed of PAN dissolved in dimethyl sulfoxide, with water to form a precursor solution.

17. The method of Claim 16, additionally includes the step of applying the mixture uniformly on the carbon paper until the paper is

- 10 -

saturated, whereafter the saturated carbon paper is cured by freezing and freeze-drying, and is the pyrolyzed to form flat, thin carbon foam.

18. A flat, thin carbon foam, produced by the method of Claim 1.

19. The carbon foam of Claim 18, having a density in the range of 0.15 to 1.0g/cc, and a thickness in the range of 1 to 40 mil.

20. A method for making a thin carbon foam electrode, comprising:  
forming a precursor solution by dissolving PAN in a solvent and mixing with water to form a solution;  
applying the mixture to carbon paper until the paper is saturated;  
freezing the PAN-saturated carbon paper;  
freeze-drying the frozen, PAN-saturated carbon paper;  
pyrolyzing the freeze-dried carbon paper, whereby a flat, thin carbon foam is formed having a density of 0.15 to 1g/cc; and  
forming the carbon foam into an electrode.

21. The method of Claim 20, wherein 12% PAN is dissolved in a solvent composed of 97% dimethyl sulfoxide and 3% water; wherein the carbon paper has a thickness of about 125 microns and density of about 0.15g/cc; wherein the PAN-saturated carbon paper is frozen at a temperature of about -5 to -150°C; wherein the frozen, PAN-saturated carbon paper is freeze-dried at about -6°C with pressure ramping from about 300 milli Torr to 0 milli Torr; and wherein the freeze-dried carbon paper is pyrolyzed in an inert atmosphere at a temperature of about 1050°C, whereby the thin carbon foam has a density of about 0.27g/cc.

22. The method of Claim 20, wherein the density of the thus formed thin carbon foam is increased by either re-saturating the sol-saturated carbon paper after curing or after pyrolysis, and then pyrolyzing again.

23. The method of Claim 20, wherein pyrolyzing is carried out in an atmosphere selected from the group of inert and oxidizing atmospheres.

24. A method for making a thin carbon foam electrode comprising:  
forming an RF precursor solution by reacting resorcinol powder, formaldehyde, deionized water and 0.5M Na<sub>2</sub>CO<sub>3</sub>;  
infiltrating carbon paper into the thus formed RF precursor solution;  
curing the carbon paper/RF solution;  
exchanging water in the carbon paper/RF with acetone;  
evaporating the acetone to form an RF/carbon paper; and  
pyrolyzing in a selected atmosphere.
25. The method of Claim 24, wherein pyrolyzing is carried out in either an inert or oxidizing atmosphere.
26. The method of Claim 24, wherein evaporation of the acetone is carried out by air drying or by subcritical or supercritical drying.
27. The method of Claim 24, additionally including doping of the thin carbon foam to provide improved electrochemical performance.
28. The method of Claim 24, additionally including doping the thin carbon foam with dopant selected from the group of lithium, phosphorous, aluminum, boron and arsenic.
29. The method of Claim 1, wherein the highly porous materials is selected from the group of membranes, papers, fabrics, felts, carbon fibers, cloths and metal fibers and powders composed of nickel, titanium, stainless steel and aluminum, and mixtures thereof.

# INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US94/09547

## A. CLASSIFICATION OF SUBJECT MATTER

IPC(6) : C01B 31/02 ; H01M 4/04  
US CL : 423/445R ; 429/218,245 ; 264/29.1,29.5  
According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 423/445R ; 429/218,245 ; 264/29.1,29.5

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US, A 5,208,003 (Simandl et al.) 04 May 1993 col. 10	1,2,11,12, 15,18,19
Y	US, A 3,927,186 (Vinton et al.) 16 December 1975 col. 4	1,2,11,12, 15,18,19
Y	US, A 3,922,334 (Marek et al.) 25 November 1975 col. 3	1,2,11,12, 15,18,19

☐ Further documents are listed in the continuation of Box C. ☐ See patent family annex.

* Special categories of cited documents:	T	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
*A* document defining the general state of the art which is not considered to be of particular relevance	X	document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
*E* earlier document published on or after the international filing date	Y	document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
*L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	&	document member of the same patent family
*O* document referring to an oral disclosure, use, exhibition or other means		
*P* document published prior to the international filing date but later than the priority date claimed		

Date of the actual completion of the international search  
19 OCTOBER 1994

Date of mailing of the international search report  
29 NOV 1994

Name and mailing address of the ISA/US  
Commissioner of Patents and Trademarks  
Box PCT  
Washington, D.C. 20231

Authorized officer  
Stuart Hendrickson

Facsimile No. (703) 305-3230

Telephone No. (703) 308-2539